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EFFICIENT SYNTHESIS OF (R)-5-(2-HYDROXYETHYL)-2(5H)-FURANONE FROM (R)-MALIC ACID

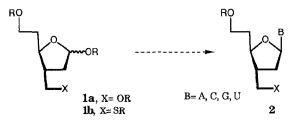
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Abstract. The title compound has been synthesized in five steps, 48% overall yield, from (R)-malic acid.

Current endeavours in these laboratories require derivatives of the branched sugars Derythro-2,3,5-trideoxy-3C-hydroxymethyl (and mercaptomethyl)-hexofuranose (1) as chiral building blocks for the synthesis of analogs of nucleosides (2).^{2,3}



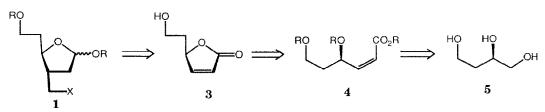
The branched-chain sugars⁴ of type **1a** or **1b** could be synthesized through a Michael addition of an hydroxymethyl⁵ or mercaptomethyl⁶ synthetic equvalent to a derivative of the butenolide **3** (scheme 1). This compound has previously been synthesized in racemic form from trans-3-hexenedioic acid, and converted to the pheromone (\pm)-eldanolide⁷. The enantiomer of compound **3** has been prepared in a very low yield by Labelle and Guindon⁸ as a side product in the studies of the mechanism of the iodoethenfication reaction of 4,6-dihydroxy-2-hexenoic acid. While the work described here was in progress, Suemune el *al.*⁹ reported the synthesis of **3** in 8 steps (6.5% overall yield) from 1,4-cyclohexadiene, using a lipase-mediated kinetic resolution as key step. Because this sequence is not entirely satisfactory for our purposes, we have developed a new route to compound **3**.

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In this paper we report the synthesis of the title compound (3) in 5 steps and 48% overall yield from commercially avalable (R)-malic acid. The synthesis of 3 also constitues a formal synthesis of natural (+)-eldanolide⁷.

Our retrosynthetic plan is shown in *Scheme 1*.

Scheme 1

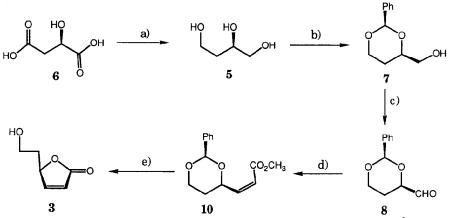


The sense of the chirality in compound **3** is the same as in (R)-1,2,4-butanetriol (**5**). To convert **5** to the butenolide **3**, it is necessary to protect selectively the hydroxy groups of C-2 and C-4 and to elongate the chain to a (Z)- α , β -unsaturated ester (compound **4**) These two goals could be achieved taking advantage of the known tendency of polyhydroxy-compounds to give 6-membered cyclic acetals when reacted with benzaldehyde under thermodinamically controlled conditions¹⁰, and the Z-stereoselective Wittig reaction between α -alkoxy aldehydes and methoxycarbonylmethylenetriphenylphosphorane (**9**, *Scheme 2*) in methanol¹¹.

These two objectives were accomplished. The overall synthetic sequence is indicated in *Scheme 2*. Reduction of (R)-malic acid (6) with borane-dimethyl sulfide complex afforded (R)-1,2,4butanetriol in 97% yield after chromatography.¹² The triol (6.3 g, 60 mmol) was refluxed with a slight excess of benzaldehyde and a catalytic amount of p-toluenesulfonic acid in toluene solution under a Soxhlet extractor charged with 4 Å molecular sieves¹³ to give a 87% yield of the six membered ring acetal (7)¹⁴, along with a trace amount of its epimer (the trans-dioxane) and a small amount of the regioisomeric dioxolanes (as a *ca.* 1:1 mixture of cis- and transdiastereoisomers, overall yield of dioxolanes: 10%).

Compound 7, after chromatographic purification, was summited to the conditions of the Swern oxidation¹⁵ to give the aldehyde 8, which was dissolved in dry methanol and reacted with the phosphorane 9 to afford a readily separable mixture of 10 and its E-stereoisomer in a 92.8 ratio in 62% overall yield from 7. Compound 10 was treated with acetic acid/water at room temperature to give the target molecule 3 in quantitative yield





a) BH3.SMe2, B(OMe3), THF, 0° to rt., 97%, b) PhCHO (1.15 eq.), TsOH (0.015 eq.), toluene, 4 Å molec. sieves, reflux, 87%; c) (COCl2), DMSO, CH2Cl2, -60°; Et3N, -60° to rt; d) PH3P=CHCO2CH3 (9) (1.5 eq.), CH3OH, -70° to rt, 57% (2 steps); e) AcOH/H2O (4·1, v/v), rt, 99%.

The synthesis reported above can be scaled up (starting from 30 g of (R)-malic acid) without loss in yield, and it provides multigram amounts of the butenolide **3**, which can be a useful chiral building block for the synthesis of branched-chain sugars as well as in the synthesis of biologically active natural products such as the cytotoxic alkaloid (+)-retronecine¹⁶, and pheromones¹⁷ related to (+)-eldanolide. Work along these lines is in progress.^{18,19}

REFERENCES AND NOTES.

- 1.- Permanent adress: Instituto de Quimica Organica General-C.S.I.C., C/ Juan de la Cierva, 3, 28006 Madrid, Spain.
- 2.- Schneider, K.C.; Benner, S.A.; Tetrahedron Lett. (1990) 31, 335.
- 3.- Schneider, K.C.; Ph. D. Dissertation E.T.H. (1989).
- 4.- For a recent example of the synthesis of C-branched sugars in furanose form, see Ohmori, J.; Shiotani, T.; Mitsunobu, O.; Chem. Lett. (1990) 303.
- 5.- Johnson, C.R.; Medich, J.R.; J. Org. Chem. (1988) 53, 4131; and references cited therein.
- 6.- Block, E.; Aslam, M; J. Am. Chem. Soc. (1985) 107, 6729.
- Hizuka, M; Hayashi, N.; Kamashita, T.; Suemune, H; Sakai, K.; *Chem. Pharm. Bull.* (1988) 36, 1550.
- 8.- Labelle, M., Guindon, Y.; J. Am. Chem. Soc. (1989) 111, 2204.
- 9.- Suemune, H.; Hizuka, M.; Kamashita, T.; Sakai, K.; Chem. Pharm. Bull. (1989) 37, 1379.

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- 10.- Foster, A.B.; in "*The Carbohydrates. Chemistry and Biochemistry*", ed.: Pigman, W.; Horton, D; Academic Press, New York, 1972, Vol. IA, pp 391-402.
- 11.- Valverde, S.; Martin-Lomas, M.; Herradon, B.; Garcia-Ochoa, S.; *Tetrahedron* (1987) 43, 1895.
- 12.- Hanessian, S.; Ugolini, A.; Dube, D.; Glamyan, A.; Can. J. Chem. (1984) 62, 2146.
- 4 Å molecular sieves are very efficient for removing water in acetalization reactions Herradon, B.; unpublished results; see also: Herradon, B.; Seebach, D.; *Helv. Chim. Acta* (1989) **72**, 690.
- 14 Compound 8 has been synthesized in 8 steps from (R,R)-dimethyl tartrate; Hungerbühler,
 E.; Seebach, D.; Wasmuth, D.; *Helv. Chim. Acta* (1981) 64, 1467.
- 15- Omura, K.; Swern, D.; Tetrahedron (1978) 34, 1651.
- 16.- Kametani, T; Yukawa, H; Honda, T.; J. Chem. Soc., Perkin 1 (1990) 571.
- 17.- Mori, K.; en "*The Total Synthesis of Natural Products*", ed.: ApSimon, J.; Wiley-Intersience, New-York (1981) Vol. 4, pp 1-183.
- 18.-All the compounds reported in this paper show satisfactory spectroscopic data. Some selected analytical data are: 7 (thick oil): $[\alpha]_{D} = -9.4$ (CHCl₃, c=1.2); ¹H-NMR (CDCl₃): $\delta =$ 7.53-7.33 (m, 5H), 5.56 (s, 1H), 4.31 (ddd, 1.3, 5.2, 11.4, 1H), 4.09-3.92 (m, 2H), 3.72-3.65 (m, 2H), 2.10 (broad s, exchange with D₂O, 1H), 2.04-1.83 (m, 1H), 1.51-1.41 (m, 1H) ppm.; ¹³C-NMR (CDCl₃): δ = 138.4 (s), 128.9 (d), 128.4 (2C, d), 126.1 (2C, d), 101.3 (d), 77.6 (d), 66.6 (t), 65.6 (t), 26.8 (t) ppm. 8 (thick oil): ¹H-NMR (CDCl₃): δ = 9.73 (s, 1H), 7.57-7.36 (m, 5H), 5.61 (s, 1H), 4.40-4.32 (m, 2H), 4.01 (dt, 2.9, 11.7, 1H), 2.06-1.76 (m, 2H) ppm.; IR (CHCl₃): v= 1705 cm-1; MS: m/z= 193 (10), 192 (M+, 65), 191 (76), 163 (20), 122 (73), 105 (100). **10**: m p.= 68-70°; $[\alpha]_{D} = -63.2$ (CHCl₃, c=1.2); ¹H-NMR (CDCl₃): $\delta = 7.53$ -7.47 (m, 2H), 7.40-7.32 (m, 3H), 6.37 (dd, 7.1, 11.7, 1H), 5.84 (dd, 1 4, 11.7, 1H), 5.62 (s, 1H), 5.53 (m, 1H), 4.30 (m, 1H), 4.11 (dt, 3.3, 11.4, 1H), 3.75 (s, 3H), 1.95-1.76 (m, 2H). **3** (oil): $[\alpha]_D = -48.2$ (CHCl₃, c=2.2) (lit.⁹: -46.4); ¹H-NMR (CDCl₃): $\delta = 7.57$ (dd, 1.5, 5.7, 1H), 6.12 (dd, 2.0, 5.7, 1H), 5.28-5.24 (m, 1H), 3.91-3.69 (m, 2H), 2.10-2.02 (m, 1H), 1.90-1.82 (m, and broad s, 2H) ppm.; ¹³C-NMR (CDCl₃): δ= 173.7 (s), 157.7 (d), 120.8 (d), 81.4 (d), 58.3 (t), 36.0 (t) ppm.; IR (CHCl₃): v= 3480, 1755, 1600, 1165, 1105, 1065, 1050, 815 cm-1; MS: m/z = 129 (24), 128 (M⁺, 24), 110 (68), 99 (67), 97 (61), 82 (100).
- 19.- Financial support from the Swiss National Foundation and from the Spanish Ministry of Education is gratefully acknowledged. I warmly thank Professor Steven A. Benner for helpful discussions.